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K19U 2460

Reg. No.:....

Name :

III Semester B.Sc. Degree (CBCSS-Reg./Sup./Imp.) Examination, November - 2019

(2014 Admn. Onwards)

COMPLEMENTARY COURSE IN CHEMISTRY 3C03 CHE(BS): CHEMISTRY (FOR BIOLOGICAL SCIENCES)

Time: 3 Hours

Max. Marks: 32

SECTION-A

Answer All questions. Each question carries 1 mark.

 $(5 \times 1 = 5)$

- What is meant by isolated systems? 1.
- Define the term conformation. 2.
- Name the functional group in 3.
 - Carboxylic acid a)
 - b) Amide
- What is meant by chirality ? 4.
- What are chelating ligands?

SECTION-B

Answer any Four questions. Each question carries 2 marks. $(4 \times 2 = 8)$

- Name the following 6.
 - Zn₂[Fe(CN)₆] a)
 - [Cr(NH₃)₆] Cl₃ b)
- Predict the products $CH_3 CH = CH_2 + HBr \rightarrow$ 7.
- 8. What is meant by heterolysis? Give one example.
- What are free radicals? Give any two reaction in which they are formed.
- 10. What are isochoric and isobaric processes?



11. The boiling point of diethyl ether is 35°C. Its heat of vaporization at its boiling point is 27.2 KJ/mole. Calculate entropy of vaporization.

SECTION-C

Answer any Three questions. Each question carries 3 marks. (3×3=9)

- 12. Write down Gibbs Helmholtz equation. What are the criterion for
- 13. Discuss Werners theory of coordination.
- 14. Discuss the optical isomerism of tartaric acid.
- 15. Explain peroxide effect with a suitable example.
- 16. Give an account of formaldehyde based plastics.

SECTION-D

Answer any **Two** questions. Each question carries **5** marks. $(2 \times 5 = 10)$

- 17. Discuss the factors affecting stability of complexes.
- 18. Explain with mechanism the various electrophilic substitution reactions **19**. a)
- Derive a relation between Cp and Cv.
 - State and explain second law of thermodynamics. b)
- 20. Write notes on
 - co polymers a)
 - biodegradable polymers. b)