

M 7158

Reg. No. : iii) According to Debye's theory of specific heat at night

Name :

Another beam is obtasped in the first code after V Semester B.Sc. Degree (CCSS – Reg./Supple./Imp.)

Examination, November 2014 Define coordination. CORE COURSE IN PHYSICS **5B08 PHY : Physics of Solids** te the mean free time in copper at 20°C, assuming one free electron/

0.97 A is obtained in the tothis option of the stration

Time : 3 Hours Max. Weightage : 30 $n = 8.46 \times 10^{28}$ electron/m³ p = 1.673 × 10⁻⁸ Ω m.

Answer any six. Each question cames A TRAPOR O

Answer all questions. Each bunch carries a weightage of 1 : 9900 meretible dutail .2

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i) Sodium has body centred packing. If the distance between two nearest atom 1.

is 3.7 Å, then lattice parameter is a polonit pappage ad agem up ob tertword.

a) 2.3 Å b) 3.9 Å c) 4.3 Å de bloch d) 4.8 Å de bloch avid a ii) A crystal that is transparent to light is due to aslo to experience out yas tall . a) lonic bonding b) Metallic bonding c) Covalent bonding Br Whatlare phonor d) Vander Waal's bonding

- iii) The co-ordination number for simple cubic crystal is g What is Meissner elle
- a) 4 b) 6 c) 8 toe d) 12 cardsact eniled or iv) The number of atom per unit cell of a bcc crystal is

a) 1 b) 2 c) 4

- i) Bragg's law is 2.
 - a) d sin $\theta = 2\lambda$ b) $2d\cos\theta = \lambda$
 - c) $2d \sin \theta = n\lambda$ d) $d \tan \theta = 2\lambda$
 - ii) When temperature of a metal increases, then its mean free path
 - a) decreases b) increases
 - c) constant

d) first increase and then decrease

Answer any nine questions. Each carries a we

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iii) According to Debye's theory of specific heat at high temperature specific heat is proportional to

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- a) T b) T² c) T³ d) Independent of T
- iv) BCS theory relating to

Max. Weightage: 30

- a) Interference b) Superconductivity
- c) Specific heat abilities to easied) None of the above

$(2 \times 1 = 2)$

(6×1=6)

PART-B

Answer any six. Each question carries a weightage of 1.

- 3. List the different type of bond with suitable example.
- 4. What is Miller indices ? ded expected entry product between budged expected.
- 5. What do you mean by packing fraction ?
- 6. Give the principle of rotating crystal method.
- 7. List any two drawbacks of classical of theory of free electron model.
- 8. What are phonons ? Give one property.
- 9. What is Meissner effect ?
- 10. Define Josephson's effect.

PART-C

III) The co-ordination number for simple

Answer any nine questions. Each carries a weightage of 2 :

- 11. Distinguish between Ionic bond and covalent bond with example.
- 12. The Bragg angle corresponding to the first order reflection from (1,1,1) planes in

a crystal is 30° when X-rays of wavelength 1.75 Å are used. calculate the interatomic spacing.

13. Derive Bragg's law and give its importance.

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14. An X-ray beam of wavelength 0.97 A is obtained in the third order after reflection at 60° from the crystal plane. Another beam is obtained in the first code after reflection at 30°. Find the wavelength of second X-ray beam.

 15. Define coordination number and lattice constant of a crystal. Explain how lattice constant in alkali halide crystals was calculated.

16. Calculate the mean free time in copper at 20°C, assuming one free electron/ copper atom. Assuming that the average speed of the free electrons is about 10^6 m/s, estimate the mean free path. n = 8.48×10^{28} electron/m³ p = $1.673 \times 10^{-8} \Omega$ m.

17. Discuss Einstein's specific heat formula and explain its limitations.

- 18. What do you understand by sp. heat of solids ? How the concept of phonons be explained ?
- 19. Explain with suitable example Type I and Type II superconductors.
- 20. Distinguish between DC Josephson effect and ac Josephson effect.
- 21. Discuss the phenomenon of thermal conductivity due to electron.
- 22. Obtain an expression for Debye's frequency.

 $(9 \times 2 = 18)$

PART-D

Answer any one. Each question carries 4 weightage.

- 23. Discuss Laue's principle of X-ray diffraction and obtain the diffraction condition for a simple cubic lattice. What is Laue spots ?
- 24. Briefly explain the salient features of BCS theory. Describe one experimental evidence for the existence of energy gap. (1×4=4)